

Chemistry CP I  
Naming Test Part 1

Name: \_\_\_\_\_

Please write the proper formula indicated by the following names. 2 points each.

1. barium fluoride \_\_\_\_\_
2. chromium (III) oxide \_\_\_\_\_
3. indium oxide \_\_\_\_\_
4. lead (IV) hydride \_\_\_\_\_
5. sodium iodide \_\_\_\_\_
6. calcium dichromate \_\_\_\_\_
7. iron (III) hydroxide \_\_\_\_\_
8. calcium chlorate \_\_\_\_\_
9. iron (II) bromide octahydrate \_\_\_\_\_
10. sodium sulfite heptahydrate \_\_\_\_\_
11. ammonium hydrogen carbonate \_\_\_\_\_
12. nickel (II) phosphate \_\_\_\_\_
13. beryllium nitrite \_\_\_\_\_
14. magnesium dihydrogen phosphate \_\_\_\_\_
15. nitric acid \_\_\_\_\_
16. chlorous acid \_\_\_\_\_
17. hydrosulfuric acid \_\_\_\_\_
18. dinitrogen tetroxide \_\_\_\_\_
19. triphosphorus pentoxide \_\_\_\_\_
20. carbon monoxide \_\_\_\_\_
21. lithium carbonate \_\_\_\_\_
22. chromium (III) peroxide \_\_\_\_\_
23. phosphoric acid \_\_\_\_\_
24. carbon tetrachloride \_\_\_\_\_
25. ammonia \_\_\_\_\_

Chemistry CP I  
Naming Test Part 2

Name: \_\_\_\_\_

Please write the proper name indicated by the following formulas. 2 points each.

26.  $\text{HBr (aq)}$  \_\_\_\_\_

27.  $\text{HNO}_2 \text{ (aq)}$  \_\_\_\_\_

28.  $\text{H}_2\text{SO}_4 \text{ (aq)}$  \_\_\_\_\_

29.  $\text{Al}_2\text{S}_3$  \_\_\_\_\_

30.  $\text{SnO}_2$  \_\_\_\_\_

31.  $\text{Pb}_3\text{P}_2$  \_\_\_\_\_

32.  $\text{NiO}$  \_\_\_\_\_

33.  $\text{Na}_2\text{S}$  \_\_\_\_\_

34.  $\text{MgCl}_2$  \_\_\_\_\_

35.  $\text{LiCN}$  \_\_\_\_\_

36.  $\text{MnHCO}_3$  \_\_\_\_\_

37.  $\text{NH}_4\text{OH}$  \_\_\_\_\_

38.  $\text{BaCrO}_4$  \_\_\_\_\_

39.  $\text{Na C}_2\text{H}_3\text{O}_2$  \_\_\_\_\_

40.  $\text{Li}_2\text{SO}_4 \bullet 8 \text{ H}_2\text{O}$  \_\_\_\_\_

41.  $\text{CoF}_3 \bullet 3 \text{ H}_2\text{O}$  \_\_\_\_\_

42.  $\text{MgSO}_3 \bullet 4 \text{ H}_2\text{O}$  \_\_\_\_\_

43.  $\text{S}_2\text{O}_2$  \_\_\_\_\_

44.  $\text{B}_3\text{O}_9$  \_\_\_\_\_

45.  $\text{XeF}_6$  \_\_\_\_\_

46.  $\text{Li}_2\text{Cr}_2\text{O}_7$  \_\_\_\_\_

47.  $\text{NaNO}_3$  \_\_\_\_\_

48.  $\text{H}_2\text{SO}_3 \text{ (aq)}$  \_\_\_\_\_

49.  $\text{Ni}(\text{ClO}_4)_3$  \_\_\_\_\_

50.  $\text{Cu}_2\text{S} \bullet 6 \text{ H}_2\text{O}$  \_\_\_\_\_